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<p>(54) Title: ANALYTE DETECTION SYSTEMS</p> <p>(57) Abstract</p> <p>The present invention is directed to improved analyte systems which are able to compensate for interfering signals and to provide accurate detection and/or quantification of analytes in a sample. In particular, the invention describes new methods, systems, and devices for indirectly determining the presence of one or more analytes in a multi-component medium.</p> <div data-bbox="797 1129 1411 1585"><p>The graph illustrates the absorbance of two compounds, Compound A and Compound B, across a range of wavelengths. The x-axis represents wavelength, with specific points 'a' and 'b' marked. The y-axis represents absorbance. Compound A shows two distinct peaks: a primary peak at wavelength 'a' (labeled Aa, Aa', Ha) and a secondary peak at wavelength 'b' (labeled Ab). Compound B also shows two peaks: a primary peak at wavelength 'a' (labeled Bb) and a secondary peak at wavelength 'b' (labeled Bb'). The peaks for both compounds are shown as overlapping curves, with dashed lines indicating the individual peak positions and heights.</p></div>		

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## ANALYTE DETECTION SYSTEMS

### FIELD OF INVENTION

This invention relates generally to analyte detection systems, and in particular to methods, systems and devices for indirectly determining the presence of one or more  
5 analytes in a multi-component medium. The invention is particularly useful in detecting the presence and/or concentration of glucose in a blood sample.

### BACKGROUND OF THE INVENTION

The detection and quantification of chemical and biochemical components in  
10 aqueous fluids, in particular biological fluids, such as, whole blood and urine and biological fluid derivatives, such as, blood serum and blood plasma, is of ever-increasing importance. Important applications exist in medial diagnosis and treatment and in the quantification of exposure to therapeutic drugs, intoxicants, hazardous chemicals and the like. In some instances, the amount of materials being determined are either so small or so  
15 difficult to precisely determine that the apparatus employed is complicated and useful only to skilled laboratory personnel. In this case, the results are generally not available for hours or days after sampling. These apparati are not for suitable for operation by patients who often need to perform such tests routinely, quickly and reproducibly outside of a laboratory setting with rapid or immediate information display.

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Devices and methods are generally known for detecting analytes, such as glucose, in a blood sample. For example, one common medical test is the measurement of blood glucose levels by diabetics. Current diabetes maintenance programs encourage patients to

measure their blood glucose level from two to seven times a day, depending on the nature and severity of the disease. Based on the observed pattern in the measured glucose levels, the patient and physician make adjustments in diet, exercise and insulin intake to better manage the disease. Clearly, the accuracy and immediacy of the results of such tests is  
5 paramount.

Thus, the art is replete with examples of methods for detecting glucose in a blood sample. For example, U.S. Patent No. 5,179,005 to Phillips et al. discloses a method for determining the presence of an analyte in a fluid. In this method, a MBTH-DMAB (3-  
10 methyl-2-benzothiazolinone hydrazone hydrochloride and 3-dimethylaminobenzoic acid) dye forms a chromophore when reacted with an analyte such as, for example, glucose. This dye absorbs at approximately 635 nm but does not absorb to any significant extent at 700 nm. Because blood contains background materials which absorb at both 635 nm and 700 nm, this method requires that a relationship between absorbants at 635 nm and 700 nm  
15 due to blood contaminants be determined by measuring the absorbance of blood samples with 0 mg/dl glucose over a wide range of blood colors. Such a color range was constructed by varying the hematocrit of the blood which results in substantially linear relationships. From these relationships, contaminating absorbance readings at 700 nm are normalized to the equivalent contaminating absorbant readings at 635 nm. This  
20 normalization was expressed as  $K/S - 15n = (K/S - 15 \times 1.54) - 0.133$ . Using this relationship, as well as others, contaminating absorbencies at 635 nm were accounted for and subtracted out to give a true glucose concentration. Such a method is limited, however, because the selected dye must absorb at a wavelength that is different from the

wavelength at which red blood cells absorb. Furthermore, the need to construct a standard for the background materials in blood is time consuming and expensive.

5 A similar test for glucose is disclosed by U.S. Patent No. 5,426,032 to Phillips et al. which describes a no-wipe whole blood glucose test strip. This test strip is adapted for use in a reflectance reading apparatus that is capable of measuring reflectance at two different wavelengths. This patent is similar to Phillips' earlier issued '005 patent because a background absorbance reading must be taken at 700 nm while the reflectance of the chromophore is read at 635 nm. Thus, in the '032 patent when measuring analytes  
10 contained in whole blood, readings are taken at two different wavelengths. The reading at one wavelength, e.g. 700 nm, is used to subtract out the background interference caused by, for example, hematocrit, blood oxygenation, and other variables which may affect the result. The method of this patent, however, is limited in that the signal-producing reagent must be absorbed at a wavelength other than a wavelength at which the assay medium  
15 substantially absorbs. Furthermore, this method is unable to correct for contaminants and/or other analytes which absorb at the chromophore's wavelength, i.e., 635 nm, and at the same time the chromophore produced by the analyte also absorbs at a second wavelength, i.e., 700 nm.

20 In a slightly different approach to correcting for contaminating signals during analyte detection in a sample, U.S. Patent No. 5,453,360 to Yu describes an oxidative coupling dye for photometric quantitative analysis of such analytes. In particular, a dye couple is described which includes 3-methyl-2-benzothiazolinone hydrazone (NBTH) and

8-aniline-1-naphthalenesulfonate (ANS). This dye couple is used as an indicator in a reaction cascade that produces a strong oxidizing agent, such as, hydrogen peroxide. This strong oxidizing agent then reacts with the dye couple to produce a blue dye reaction product. The NBTH-ANS dye couple exhibits strong and flat spectral absorption at about 5 600 to 650 nm. This region of absorbance is free of blood color interference.

Accordingly, glucose and other analytes that react with an oxidase enzyme to produce the above-described strong oxidizing agent can be accurately measured without much optic calibration. This patent is limited because the dye couple must absorb in an area of the reflectance spectrum between 600 and 650 nm which is outside the region of blood color 10 interference or the dye couple must be used in the absence of red blood cells so that it would not be subject to interference by the color of the blood. More generally, this patent suffers from the drawback that unknown components in a sample which happen to absorb at the detection wavelength will cause the analyte measurements to be over-estimated.

15 Similarly, U.S. Patent No. 5,389,524 to Larsen et al. describes a method and an apparatus for quantitatively monitoring a chemical component dissolved in a liquid medium. This patent describes taking two measurements of a system at the same wavelength. The absorbance of a colored reaction product is determined by subtracting an end-point absorbance from a background absorbance. Thus, this patent takes two 20 absorbance measurements at the same wavelength and is susceptible to erroneous results if one or more analytes or interfering substances happen to absorb at the same wavelength that the two absorbance measurements are taken.

Non-glucose analyte detection systems are also well known in the art. For example, U.S. Patent No. 5,204,242 to Junius-Comer describes the use of a substituted phenol with high stability and low non-specific reactivity as a coupling component used in a colormetric process for detecting oxidative coupling reactions, such as the determination of enzymatic creatinine levels. In this method, a reference measurement is made at 700 nm before the reaction starts; while the extinction increase is measured at 546 nm between the sixth and tenth minute of reaction. This method is limited by its assumption that no interfering component absorbs at the detection wavelength, i.e., 546 nm.

It is also known in the art to use fluorescent or other detection schemes for determining the presence and/or quantity of an analyte in a sample. For example, U.S. Patent No. 5,527,684 to Mabile et al. describes a method of measuring the luminescence emitted in a luminescent assay. This method includes forming a reaction mixture by contacting a sample with an internal reference compound that emits at a first wavelength and a tracer compound that emits at a second wavelength. The sample is then irradiated at a single excitation wavelength. The tracer is detected by measuring the emitted luminescence at the second wavelength. The internal reference is detected by measuring the emitted luminescence at the first wavelength. Using the luminescence emitted at the first wavelength by the internal reference, a correction is made to the luminescence emitted at the second wavelength by the tracer. In this way, the analyte in question can be quantified or detected.

In summary, the prior art methods cited above all suffer from the drawback that

they are susceptible to over-estimating and/or providing false positive results if a contaminating component of a sample happens to absorb at the detection wavelength.

Thus, there is a need for an improved analyte detection system which is able to compensate for interfering absorbencies and to provide accurate detection and/or quantification of analytes in a sample. In particular, there is a need for quick, cost-effective, accurate and easy-to-use methods and systems for indirectly determining the presence and/or quantity of one or more analytes in a system. The present invention is directed to meeting these and other needs.

## 10 SUMMARY OF THE INVENTION

In one embodiment of the present invention, there is provided method for indirectly determining the presence of one or more analytes in a multi-component medium. This method includes reacting the multi-component medium with one or more signal-producing reagents to form one or more signal producing molecules which produce characteristic spectra in response to one or more stimuli in that the ratios of these signals produced from the same molecules are constants and are characteristic of the molecule. A stimulus or stimuli is then provided or self-produced by the signal producing molecule for exciting the multi-component medium whereby the number of stimuli provided or self-produced is equal to or greater than the sum of the number of the analytes and any interfering components in the multi-component medium, provided that when there is more than one analyte or interfering component in the multi-component medium that the stimuli provided or self-produced to correct the interference are different from a stimulus for measuring the analyte, and the interference-correcting signals at these stimuli do not



overlap each other. This method further includes detecting at least a first and second signal of one or more analytes generated in response to the stimulus or stimuli. The presence and/or concentration of one or more of the analytes is determined by applying mathematical formulas to at least the first and second signals.

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In another embodiment of the present invention, there is provided an analyte detection system for determining the presence and/or concentration of glucose in a blood sample which is in contact with a porous substrate having a blood application surface and an analyte detection surface thereon, wherein the presence and/or concentration of glucose cannot be directly determined. This system includes a means for introducing the blood sample onto the blood application surface of the substrate. The sample is then allowed to migrate through the porous substrate to the analyte detection surface. The detection surface includes a glucose-specific reagent and indicator disposed thereon. A signal producing composition is allowed to form on the analyte detection surface through the reaction of the glucose-specific reagent, the indicator and the glucose present in the blood sample. The sample is then stimulated with one or more stimuli used to correct for interference generated by another analyte or interfering component. The signals produced in response thereto are then detected. Such detection is based on the requirement that the number of interference-correcting stimuli required is equal to or greater than the sum of the number of analytes in the sample and any interfering components therein, provided that the stimuli are different from a stimulus for measuring the glucose, and that these interference-correcting stimuli do not overlap each other. The presence and/or concentration of the glucose is determined by applying mathematical formulas as further

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described herein to the produced signal(s).

In a further embodiment of the present invention, there is provided a process for detecting one or more analytes in a multi-component system, wherein the ability to  
5 directly detect one or more analytes is masked. This process includes forming a signal producing composition by reacting an analyte with a reagent. The system is then excited with one or more stimuli to produce an interference-correcting signal from the signal producing composition, whereby the number of stimuli provided is equal to or greater than the sum of the number of analytes and any interfering components in the multi-component  
10 medium, provided that the interference-correcting stimuli are different from a stimulus for measuring the analyte, and that these interference-correcting signals at these stimuli do not overlap each other. The signal or signals are then measured and the presence and/or concentration of one or more of the analytes is determined by applying mathematical formulas to the signal or signals. These mathematical formulas are defined as follows:

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$$\text{Concentration of Analyte} = E \times A_a$$

wherein E is an extinction coefficient and  $A_a = (A_{a'} - k \times B_{b'}) / (1 - k \times c)$  and wherein  $A_a$  is the true value for the analyte,  $A_{a'}$  is a measurable signal produced in response to a single  
20 stimulus by the signal producing composition ( $A_a$ ) and no more than one interfering component ( $B_a$ ) at the first detection point,  $B_{b'}$  is a measurable signal produced in response to the stimulus by the signal producing composition ( $A_b$ ) and no more than one interfering component ( $B_b$ ) at the second detection point,  $k$  is a constant and is also the

ratio between Bb and Ba, and  $c$  is the ratio between Aa and Ab.

In a still further embodiment of the present invention, there is provided a device for detecting the presence and/or concentration of one or more analytes in a blood sample when the analyte cannot be directly detected. This device includes a hydrophilic porous substrate with a blood application surface and an analyte detection surface. The substrate includes a polymeric matrix having pores dispersed throughout the matrix and in communication with the blood application and analyte detection surfaces. The diameter of the pores of the polymeric matrix are sufficiently large to allow red blood cells in the sample to freely migrate between the blood application and analyte detection surfaces.

An analyte-specific reagent and an indicator are dispersed on the analyte detection surface for reacting with the analyte to produce a signal producing composition. When excited, this signal producing composition produces at least two interference-correcting signals at wavelengths ( $\lambda_a$ ) and ( $\lambda_b$ ), respectively to allow for the correction of any interfering signals produced by other analytes or interfering components in the sample. The device also includes excitation means for exciting the sample on the analyte detection surface. These excitation means can be, for example, a laser, a visible light source, etc. Moreover, detection means are also provided for detecting at least two signals. The number of interference-correcting signals produced, however, must be equal to or greater than the sum of the number of the analytes in the blood sample and any interfering components therein, provided that the interference-correcting signals are different from a signal for measuring the analyte, and that the interference-correcting signals at these

stimuli do not overlap each other. The device further includes a means for determining the presence and/or concentration of the analyte by applying mathematical formulas to at least the first and second signals. These formulas are defined as follows:

5 Concentration of Analyte = ExAa

wherein E is an extinction coefficient and  $A_a = (A_a' - kx B_b') / (1 - kxc)$  and  $A_a$  is the true value for the analyte,  $A_a'$  is a measurable signal produced at  $\lambda_a$  by the signal producing composition ( $A_a$ ) and no more than one interfering component from the sample ( $B_a$ ),

10  $B_b'$  is a measurable signal produced at  $\lambda_b$  by the signal producing composition ( $A_b$ ) and no more than one interfering component from the sample ( $B_b$ ),  $k$  is a constant and is the ratio between  $B_b$  and  $B_a$ , and  $c$  is the ratio between  $A_a$  and  $A_b$ . A display means is also provided for displaying the presence and/or concentration of the analyte.

**15 DETAILED DESCRIPTION OF THE INVENTION**

The present invention is directed to a method for indirectly determining the presence of one or more analytes in a multi-component medium, wherein the presence of one or more analytes to be detected are proportional to the magnitude of a signal (Aa) as set forth in more detail hereinbelow. For purposes of the present invention, the use of the term "indirectly" means that the present method is able to detect an analyte whether or not its signal is masked by an interfering contaminant, the color of the sample, or another analyte(s).

For purposes of the present invention, an "analyte" is any composition found in a sample for which it would be useful to detect its presence and/or quantity. More particularly, the analytes of the present invention can be any medically significant composition for which the presence or quantity thereof in a body is important to ascertain.

5 Non-limiting examples of such analytes include glucose, creatinine, cholesterol, uric acid, methanol, ethanol, formaldehyde, glycerol-3-phosphate. In addition, kinetic assays, such as those used to detect or measure enzymes can also be used in accordance with the present invention.

10 The analyte or analytes of the present invention are detected in a multi-component medium. For purposes of the present invention, "multi-component medium" refers to both aqueous and non-aqueous solutions containing at least one analyte to be detected. The multi-component medium can also contain other analytes or interfering components which may or may not interfere with the detection of the analyte of interest. Preferably, the  
15 multi-component medium of the present invention refers to biological fluid samples, including for example, blood, urine, saliva and derivatives thereof. Derivatives of blood include, for example, serum and plasma, as well as, artificial and/or synthetic blood.

In the present invention, the multi-component medium is reacted with one or more  
20 signal producing reagents to form one or more signal producing molecules. For purposes of the present invention, the term "signal producing reagents" refers to compositions which react with one or more analytes in a multi-component medium to produce one or more reaction products. These signal producing reagents are selected with reference to the

analyte or analytes to be detected, the composition of the multi-component medium and the detection device. The selection of appropriate signal producing reagents is within the knowledge of the skilled artisan given the above-referenced variables.

5           For purposes of illustration only and not by way of limitation, the signal producing reagents of the present invention can include for example, chromogens, such as 3,3',5,5'-Tetramethylbenzidine and Trinder reagents, glucose oxidase, peroxidase, and oxygen acceptors such as, O-dianisidine, O-toluidine, O-tolidine, benzidine, 2,2'-azinodi-(3-ethylbenzthiazoline sulphonic acid-(6)), 3-methyl-2-benzothiazolinone hydrazone and  
10       N,N-dimethylaniline, phenyl and 4-aminophenazone, sulfonated 2,4-dichlorophenol and 4-aminophenazone, 3-methyl-2-benzothiazolinone hydrazone and 3-(dimethylamino)benzoic acid, 2-methoxy-4-allyl phenol and 4-aminoantipyrenedimethylaniline.

          Furthermore, the present invention contemplates the use of single or multi-signal-  
15       producing reagents, depending on the analyte or analytes to be detected. Moreover, the present invention also contemplates the use of systems of signal producing reagents which work together to produce a signal producing molecule and/or for enhancing the signal produced by such a molecule.

20           For purposes of the present invention, "signal producing molecule" or "signal producing molecules" refers to the reaction product produced from reacting one or more signal-producing reagents with the multi-component medium. Such signal producing molecules produce a measurable event that is used in the present invention to detect or

quantitate the analyte or analytes of interest. This measurable event is typically, for example, a signal which is detectable using conventional methods including, absorbance, voltage, radioactivity, nuclear magnetic resonance, temperature, current and fluorescence. Furthermore, the present invention also encompasses signal producing molecules which  
5 produce a measurable response which has a characteristic ratio of signals at different stimuli.

The signal produced by these signal producing reagents may be self-produced or provided in response to one or more stimuli. In either case, these signals usually produce a  
10 spectrum which is detectable by one of the methods set forth above. In the present invention, there is a requirement that the number of stimuli provided or self-produced must be equal to or greater than the sum of the number of analytes and any interfering components in the multi-component medium. Thus, in a simple system of glucose and water, where glucose is the analyte to be determined, the number of stimuli must be one or  
15 more. In a system where there are two analytes or one analyte and one interfering component, there must be at least two stimuli in order to properly quantitate or detect the analyte of interest. There is no theoretical limit to the number of analytes or interfering components that can be present in the multi-component medium so long as the number of stimuli provided or self-produced is equal to or greater than the sum of the analytes and  
20 interfering components therein. For purposes of the present invention, an "interfering component" is any composition present in the multi-component medium which is capable of interfering with the measurement of the analyte of interest.

In the present invention, at least a first and a second signal are detected in response to the signal generated by the signal producing molecules when the stimulus or stimuli are provided to the multi-component medium. These signals are detected using conventional technology, such as for example, spectrophotometers, fluorescence spectrophotometers, voltage detectors, etc. The selection of an appropriate device for detecting such signals is within the knowledge of one skilled in the art with reference to the analytes to be detected, the signal-producing reagents used, the signal producing molecules formed, as well as, the composition of the multi-component medium.

Once a signal is detected, the presence and/or concentration of one or more analytes is determined by applying mathematical formulas to at least the first and second signals produced. These mathematical formulas are defined according to the following formula:

$$\text{Concentration of Analyte} = E \times A_a \quad (\text{Formula I})$$

wherein E is an extinction coefficient and  $A_a = (A_a' - k \times B_b') / (1 - k \times c)$  and  $A_a$  is the true value for the analyte  $A_a'$  is a measurable signal produced in response to a stimulus by the signal producing molecule ( $A_a$ ) and no more than one of the interfering components ( $B_a$ ) at  $\lambda_a$ ,  $B_b'$  is a measurable signal produced in response to the stimulus by the signal producing molecule ( $A_b$ ) and no more than one of the interfering components ( $B_b$ ) at  $\lambda_b$ ,  $k$  is a constant and is a ratio between  $B_b$  and  $B_a$ , and  $c$  is the ratio between  $A_a$  and  $A_b$ .



With reference to Figure 1, the present invention includes a signal producing system which reacts with an analyte (Compound A) to produce a signal, such as for example a chromophore, which absorbs at a specific wavelength ( $\lambda_a$ ). The intensity of the absorbance at this wavelength is measured and is then related to the concentration of the analyte. If background or contaminant materials, i.e., other analytes or interfering components, (Compound B) also absorb at the same wavelength as the chromophore, an absorbance measurement is taken at a second wavelength ( $\lambda_b$ ), even if the chromophore also absorbs at this second wavelength ( $\lambda_b$ ).

Thus, in Figure 1,  $A_a$  is the absorbance of compound A at wavelength  $\lambda_a$ ;  $A_b$  is the absorbance of compound A at wavelength  $\lambda_b$ ;  $B_a$  is the absorbance of compound B at wavelength  $\lambda_a$ ; and  $B_b$  is the absorbance of compound B at wavelength  $\lambda_b$ . Accordingly, the total absorbance measured at wavelength  $\lambda_a$  is the sum of the absorbances of compounds A and B at wavelength  $\lambda_a$  ( $A_a' = A_a + B_a$ ). Similarly, the total absorbance measured at wavelength  $\lambda_b$  is the sum of the absorbances of compounds A and B at wavelength  $\lambda_b$  ( $B_b' = B_b + A_b$ ). Thus, in order to correct for the interference of compound B at wavelength  $\lambda_a$ , the following formula is used:

(II)  $A_a = A_a' - B_a$ . Substituting for  $B_a$ , formula (III) is derived:  $A_a = A_a' - B_a$  or  $A_a =$

$A_a' - k \times B_b$ . In formula (III) by substituting for  $B_b$  with ( $B_b' - A_b$ ), formula (IV) is

derived:  $A_a = A_a' - k \times (B_b' - A_b)$ , which is the same as formula (V):  $A_a = A_a' -$

$k \times B_b' + k \times A_b$ , where  $k$  is a constant and represents the ratio between  $B_b$  and  $B_a$ , and  $B_b'$

can be measured.  $A_b$ , however, cannot be measured directly because it is buried under  $B_b$ .

Accordingly,  $A_a'$  cannot be corrected. There is, however, always a constant ratio between

Aa and Ab just as there is always a ratio between Bb and Ba (although the ratios between Aa and Ab and Bb and Ba are not necessarily the same). Thus,  $Ab = cxAa$  which can be substituted into formulae (V) to yield the following equation: (VI)  $Aa = Aa' - kxBb' + kxcxAa$  which can be rearranged to formula (VII)  $Aa = (Aa' - kxBb')/(1-kxc)$ . With  
5 formula (VII), Ab does not have to be measured so that a true absorbance of compound A at wavelength  $\lambda_a$  can be obtained.

The present invention is based on the observation that the signal producing composition is able to produce a measurable response having a characteristic ratio of  
10 signals at two different wavelengths. This characteristic ratio of signals at two different wavelengths is the basis for the existence of the constants  $k$  and  $c$ , respectively, and for the ability of the present method to detect and/or quantitate analytes in multi-component systems.

15 Thus, by taking a second absorbance measurement, one is able to correct for the absorbance of the interfering or contaminating materials. In this way, an accurate measure is obtained for the concentration of an analyte in a particular system. Accordingly, this method allows for the quick and accurate detection and/or quantification of an analyte in a multi-component medium that is independent of the wavelength at which the absorbance  
20 of a chromophore produced when a chromogen reacts with an analyte of interest.

Moreover, as set forth above and further illustrated in Figure 2, the present method also allows for the detection of three or more analytes in a multi-component medium as

long as the number of stimuli used are equal to or greater than the sum of the analytes or interfering components in the multi-component medium.

Thus, in Figure 2, the present method is able to detect and/or quantitate the presence of an analyte when multiple analytes or interfering components, i.e., three or more, mask the signal produced by the analyte of interest because the correction of the interfering signals Ba and Ca can be accomplished by measuring Bb' and Cc' at two different wave lengths. Such corrections are made using Formula I above as shown in the following derivation:

$$\begin{aligned}
 10 \quad Aa &= Aa' - Ba - Ca \\
 &= Aa' - kbxBb - kcxCc \\
 &= Aa' - kb(Bb' - Ab) - kc(Cc' - Ac) \\
 &= Aa' - kbxBb' + kbxA b - kcxCc' + kcxA c \\
 &= Aa' - kbxBb' + kb x k d x Aa - kcxCc' + kc x k e x Aa \\
 15 \quad &\text{Rearrange, } (1 - kb x k d - kc x k e) Aa \\
 &= Aa' - kbxBb' - kcxCc' \\
 &Aa = (Aa' - kbxBb' - kcxCc') / (1 - kb x k d - kc x k e)
 \end{aligned}$$

Thus, the present method is effective at determining the presence and/or quantity of an analyte in a sample provided that there are no interfering components overlapping at a stimulus, i.e.,  $\lambda_b$  or  $\lambda_c$ , that must be used for corrections. This method is able to both detect the presence of and/or quantitate the amount of an analyte in the multi-component medium of the present invention. For example, an analyte's presence may be detected

with a percent coefficient of variance of about 30% to about 50%. Furthermore, an analyte can be quantified, i.e., its concentration may be determined, with a percent coefficient of variance of about 3% to about 7%. The coefficient of variance with respect to the detection and/or quantitation of any particular analyte will, of course, vary depending upon the measuring devices and reagents used.

In the present invention, the stimulus or stimuli used to excite the multi-component medium can be derived from any source provided that such stimuli are able to induce the signal producing molecule to produce a detectable signal. Such stimuli may include for example, radiation from the electromagnetic spectrum, current, nuclear magnetic resonance, voltage, fluorescence, radioactivity and temperature.

As described above with reference to Figures 1 and 2, multiple signals may be detected by exciting or stimulating the multi-component medium at more than one point, such as for example, at more than one excitation wavelength and subsequently measuring the corresponding absorbencies at at least first and second wavelengths  $\lambda_a$  and  $\lambda_b$  (labeled a and b in Figures 1 and 2). As set forth previously, the number of signals to be detected will be determined with reference to the number of stimuli provided and the number of analytes and interfering substances for which corrections must be made. In a preferred embodiment of the present invention, for example,  $\lambda_a$  and  $\lambda_b$  are 630 nm and 700 nm, respectively.

In another embodiment of the present invention, an analyte detection system is

provided for determining the presence and/or concentration of glucose in a blood sample which is in contact with a porous substrate having a blood application and an analyte detection surface. In this system for detecting glucose in a blood sample, the presence and/or quantity of glucose cannot be directly determined because of the presence of at least one interfering signal generated by the hemoglobin in the red blood cells of the blood sample.

The porous substrate of this embodiment includes a matrix of a hydrophilic polymer with pores dispersed throughout the matrix. These pores are in communication with both the blood application and analyte detection surfaces. Moreover, the diameter of the pores are sufficiently large to allow red blood cells in the blood sample to freely migrate between the blood application and the analyte detection surfaces.

In this system, a blood sample is introduced onto a blood application surface as set forth above. To work, the present system only requires a drop of blood from a patient. The volume of such a drop of blood can range from about 10  $\mu$ l to about 100  $\mu$ l. The blood application surface of the present system can be any porous surface which is adapted to receive blood. This surface, however, must be adapted to allow red blood cells to freely migrate between the blood application surface and the analyte detection surfaces via the pores in the hydrophilic matrix of the substrate. Accordingly, the blood application surface is in fluid contact with the analyte detection surface so that glucose, as well as other analytes, are not hindered in the movement to the analyte detection surface.

The analyte detection surface of the present invention contains a glucose specific reagent and an indicator, although other reagents and indicators may be substituted therefor to detect analytes different from glucose. The reagent and indicator may be disposed on the analyte detection surface itself or may be impregnated within the pores of the polymer matrix. The only limitation as to how the reagent and indicator are arranged on the analyte detection surface is that they must be accessible to react with the analytes, i.e., glucose, contained within the blood sample.

When the blood sample reaches the analyte detection surface of the present invention, a signal producing composition is formed through the reaction of the glucose-specific reagent, the indicator and the analyte, i.e., glucose, in the blood sample as described in detail above. The detection of any signals produced from this reaction, as well as the determination of the presence and/or quantity of glucose in a blood sample has been described above with reference to Figures 1 and 2.

The presence and/or concentration of glucose in the blood sample can be determined as previously set forth. Furthermore, one or more signals can be detected by stimulating or exciting the blood sample and taking measurements as described hereinabove. Moreover, Formula I, as set forth above, is used in the present system for determining the presence and/or concentration of glucose, or any other analyte compatible with the present invention.

As set forth above, it is preferred that  $\lambda_a$  and  $\lambda_b$  described in Formula I are 630 nm

and 700 nm, respectively. Other wavelengths, however, are also contemplated by the present invention depending upon the analyte to be detected, the sample, and the reagent and indicators used.

5           In the present system, the signal producing composition, as set forth above, may produce a measurable response having a characteristic ratio of signals at two different wavelengths. Furthermore, this ratio is the basis for the existence of constants  $k$  and  $c$ , respectively. Also as set forth above, a signal is produced by the signal producing composition which includes absorbance, voltage, radioactivity, temperature, current and  
10       fluorescence. The glucose-specific reagent may include a chromogen, such as for example 3,3',5,5'-Tetramethylbenzidine; a Trinder reagent; or any other substance capable of producing a signal detectable by the present system.

          In another embodiment of the present invention, a process for detecting one or  
15       more analytes in a multi-component system is provided in which the ability to directly detect the analytes is masked by no more than one interfering component of the system. This process includes forming a signal producing composition by reacting an analyte with a reagent, as set forth above. The process further includes exciting the system with more than one stimulus to produce signals from the signal producing composition, as previously  
20       described. The signal or signals are then measured and the presence and/or concentration of one or more of the analytes is determined according to the method set forth above.

          In a further embodiment of the present invention, a device is provided for detecting

the presence and/or concentration of one or more analytes in a blood sample when the analyte cannot be directly detected. This device includes a hydrophilic porous substrate with a blood application surface and an analyte detection surface which are described in more detail above.

5

An analyte-specific reagent and an indicator as described in more detail above are dispersed on the analyte detection surface for reacting with the analyte to produce a signal producing composition. When excited, this signal producing composition produces at least two signals at wavelengths ( $\lambda_a$ ) and ( $\lambda_b$ ), respectively. The number of signals produced, however, must be equal to or greater than the sum of the number of the analytes in the blood sample and any interfering components therein, provided that the signals are different than a signal for measuring the analyte.

10

The device also includes excitation means for exciting the sample on the analyte detection surface. Moreover, detection means are also provided for detecting at least two signals. The excitation and detection means can include spectrophotometers, fluorescent spectrophotometer, voltage detectors, etc. as previously set forth. It is within the knowledge of one skilled in the art to select appropriate excitation and detection means based on the analyte to be detected and the detection reagents used.

15

20

The device further includes a means for determining the presence and/or concentration of the analyte by applying mathematical formulas to at least the first and second signals. The formulas are defined in Formula I above. A display means is also



provided for displaying the presence and/or concentration of the analyte. Such display means can be, for example, in the form of a monitor, a computer printout including tables of data and graphical representations, a strip of material which visually cues an observer as to the presence and/or concentration of a particular analyte and the like. Such display means are well known in the art and may be selected by the skilled artisan with reference to the intended application.

The following examples are set forth to illustrate the methods of the present invention. These examples are provided for purposes of illustration only and are not intended to be limiting in any sense.

#### **EXAMPLE 1A**

##### **Preparation of a Substrate to be Coated**

A test strip according to the present invention was prepared as follows: a piece of nylon membrane (Magna, 1.2 $\mu$ m from Micron Separations, Inc. Westborough, MA) was impregnated by dipping in a 1% of DOSS solution (0.2 gm Dioctyl sulfosuccinate sodium salt in 20 gm distilled water). The nylon membrane was then suspended and allowed to dry at 65° C for 15 minutes in an oven.

#### **EXAMPLE 1B**

##### **Preparation of An Analyte Detection Composition According to the Present Invention**

An analyte detection composition, Composition A was prepared as follows: 0.6 g of microcrystalline cellulose particles was added to 12.5 g of distilled water, and stirred for 30 minutes. To 3 g of the cellulose particle solution was added 0.6 ml of 0.5 M phosphate buffer (pH 5.0). This composition was stirred for about 20 minutes. The following reagents were then added to this composition: 2.5 g of hydroxyethyl cellulose ( $7.2 \times 10^5$  M.W.) solution (0.96 g of hydroxyethyl cellulose in 36 g of 0.5 M phosphate buffer (pH 5), which was stirred overnight); 2.6 g of a vinyl acetate-ethylene copolymer (Airflex 300 from Air Products & Chemicals, Inc., Allentown, PA); 0.4 ml of Igepal CO-530 (prepared by mixing 0.2 g of Igepal CO-530 with 1.65 g of distilled water); and 0.4 ml of semicarbazide solution (0.1 g of semicarbazide dissolved in 1 g of distilled water). The resulting mixture was stirred for about 20 minutes. Then, 0.3 ml of a  $\text{CaCl}_2$  solution (1.4704 g of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  dissolved in 10 g of distilled water) was slowly added to the above-described composition with constant stirring for about 15 minutes. 0.1 ml of a BHT solution (prepared by adding 0.1 g of 2,6-di-tert-butyl-p-cresol dissolved in 0.765 g of acetone), 0.67 ml of TMB solution (0.14 g of 3,3',5,5'-tetramethylbenzidine and 0.14 g of dioctyl sulfosuccinate sodium salt dissolved in 0.8 g of acetone), 0.5 ml of 4-sulfobenzoate solution (2.4 g of 4-sulfobenzoic acid, potassium salt was mixed with 10g of distilled water, and the pH of the solution was adjusted to 5.0 with 1.75 ml of a 10 N NaOH solution. This solution was then filtered through filter paper) and 0.84 ml of enzyme solution in that order, were slowly added to this composition. The enzyme solution was prepared with 5232  $\mu$  glucose oxidase, 13397  $\mu$  horseradish peroxidase, 0.2 g of Crotein spc (Croda Inc. NJ) and 0.6 g of distilled water which were stirred and mixed well.

**EXAMPLE 1C****Coating a Substrate with an Analyte Detection Composition of the Present Invention**

5           On a glass plate, Composition A of Example 2 was applied onto the dried nylon membrane of Example 1 to a wet thickness of 6.4 ml with a glass rod whose two ends were taped to afford the desired wet coating thickness. The glass plate with the coated membrane on it was dried in an oven at 65°C for 15 minutes.

10

**EXAMPLE 1D****Comparison of Prior Art Glucose Test Strip and Test Strips of Present Invention**

15

The coated substrate of Example was 1C cut into test strips which are compatible with commercially available glucose meters, such as, One Touch™ II or Basic glucose meters.

20

The spiked blood samples with various hematocrit levels were applied to the test strips of the present invention (runs 1-8), as well as to a prior art glucose test strip. The test strips were then placed in One Touch™ glucose meters and the glucose concentrations were detected. The results of these tests are indicated in Table I below:

**Table I**

		Sample 1**	Sample 2	Sample 3**	Sample 4	Sample 5**	Sample 6
		mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose
		42% Hematocrit	52% Hematocrit	42% Hematocrit	52% Hematocrit	42% Hematocrit	52% Hematocrit
	Prior Art Strip	85	77	157	145	454	427
5	run 1	90	83	223	174	533	510
	run 2	90	84	186	191	545	531
	run 3	94	74	232	182	578	527
	run 4	85	84	196	213	516	513
	run 5			205	175	525	545
10	run 6			213	196	556	502
	run 7			208	188	541	523
	run 8			211	201	542	517
	Ave.	89.75	81.25	209.25	190	542	521
	Norm*	89.75	89.69	209.25	205.72	542	553.94

15 \*\* Samples 2, 4, and 6 of higher hematocrit were prepared from Samples 1, 3, and 5 of lower hematocrit by removing plasma therefrom.

\* Strip values of the higher 52% hematocrit samples were normalized by multiplying by a certain correction factor: (One Touch value of 42% hematocrit sample which was used to  
20 prepare the 52% hematocrit sample / value of 52% hematocrit sample). Thus, the normalized values of the higher hematocrit samples can be compared easily with those of the lower hematocrit levels.

These results indicate that the test strips of the present invention are able to determine  
25 the level of glucose present over a wide range of concentrations. Furthermore,

the results indicate that interference due to higher hematocrit levels can be easily compensated for using One Touch™ glucose meters.

### **EXAMPLE 2A**

#### **Determination of Glucose Concentration in a Blood Sample Using a Trinder Reagent**

A nylon membrane as described in Example 1A was coated with a Trinder reagent and with Composition B of the present invention using the same process set forth in Example 1C. Composition B was prepared by combining the following reagents: 0.6 g 2,5-Furandione, polymer with methoxyethene solution (prepared by adding 6 g of Gantrez AN-169 (ISP Technologies, Inc., Wayne, NJ) to 72 g of distilled water, and stirring at 80°C for two hours until the solution became clear. The solution was adjusted to pH 4.0 with 3.2 ml of 10 N NaOH solution); 1 g of a DOSS solution (The DOSS solution was prepared by mixing 0.65 g of dioctyl sulfosuccinate sodium salt and 5 ml of distilled water for 30 minutes.); 0.7 ml of 4-sulfobenzoate solution (This solution was prepared by adding 2.4027 g of 4-sulfobenzoic acid, potassium salt to 10 g of distilled water with stirring while 1.75 ml of 10 N NaOH was added to adjust the pH to 5.0. This solution was then filtered through filter paper.); 0.8 ml of Igepal CO-530 (prepared by mixing 0.03 g of a surfactant (Igepal CO-530) with 1 g of distilled water); 0.3 ml of semicarbazide (0.1 g semicarbazide dissolved in 1 g of distilled water); 1 ml of phosphate buffer (pH 5); 0.3 ml of  $\text{CaCl}_2$  (containing 1.4704 g of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  dissolved in 10 g of distilled water); and 2.375 ml of distilled water. These reagents were stirred for 30 minutes. Then, 0.84 ml of an enzyme solution was added and the composition was stirred for an additional 20

minutes. The enzyme solution was prepared with 1949  $\mu$  glucose oxidase, 1935  $\mu$  horseradish peroxidase, 0.2 g of Crotein spc (Croda Inc. NJ) and 0.6 g of distilled water which were stirred and mixed well.

5           The Trinder reagent was prepared by adding 1 g of the 2,5 Furandione, polymer with methoxyethene (Gantrez An-169 solution) to 9 g of distilled water, and adjusting the pH to 7.0 with 10 N NaOH solution while the solution was stirred. To 0.2 g of this Gantrez solution was added 24.6 mg of N-ethyl-N-(2-hydroxy-3-sulfopropyl)-3,5-dimethylaniline, sodium salt, monohydrate (MAOS), 15.2 mg of 4-aminoantipyrine, 0.6 g  
10 of distilled water. The solution was stirred to dissolve. Then 1.1 ml of ethanol was slowly added while stirring.

#### **EXAMPLE 2B**

##### **Comparison of Prior Art Glucose Test Strip and Test Strip of Present Invention**

15

Using composition B and the Trinder reagent solution and the nylon substrate described in Example 2A, the procedure set forth in Example 1D was repeated. The results of the glucose determinations are set forth below in Table II.

20

**Table II**

	Sample 1**	Sample 2	Sample 3**	Sample 4	Sample 5**	Sample 6
	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose
	42% Hematocrit	52% Hematocrit	42% Hematocrit	52% Hematocrit	42% Hematocrit	52% Hematocrit
Prior Art Strip	87	81	162	146	462	423
run 1	75	66	125	114	346	312
5 run 2	71	68	128	116	358	326
run 3	74	65	127	113	335	322
run 4	72	69	126	117	369	316
Ave.	73	67	126.5	115	352	319
Norm.*	73	71.96	126.5	127.6	352	348.41

10 \*\* Samples 2, 4, and 6 of higher hematocrit were prepared from Samples 1, 3, and 5 of lower hematocrit by removing plasma therefrom.

\* Strip values of higher 52% hematocrit sample were normalized by multiplying by a certain correction factor (One Touch value of 42% hematocrit sample which was used to prepare the 52% hematocrit sample / value of 52% hematocrit sample). Thus, the normalized values of the samples can be compared easily with those of the lower hematocrit levels.

These results indicate that test strips of the present invention are able to determine the levels of glucose present over a wide range of concentrations. Furthermore, the results indicate that interference due to higher hematocrit levels can be easily compensated for using One Touch™ glucose meters.

#### **EXAMPLE 3A**

Coating A Substrate With an Analyte Detection Composition of the Present

Invention

5 On a glass plate, a Trinder reagent solution was coated onto a piece of nylon membrane (Biodyne B, 5  $\mu\text{m}$  from Pall Biosupport Division, East Hills, NY) to a wet thickness of 6.2 ml as described in Example 1C. The coated membrane was then dried in an oven at 65°C for 20 minutes. Composition C as described in further detail below was then applied to the coated membrane in the same fashion as set forth in Example 1C.

10 Composition C was prepared by combining the following reagents: 2.4 g of a solution containing 2,5 Furandione, polymer with methoxyethene (6 g of Gantrez AN-169 in 72 g of distilled water which was stirred and heated to 80°C for two hours until it became a clear solution. Then, the solution was adjusted to pH 4.0 by 3.2 ml of 10 N NaOH solution.); 1.0 ml of 0.5 M phosphate buffer (pH 5.0); 0.65 g of DOSS solution  
15 (prepared by adding 0.65 g of dioctyl sulfosuccinate sodium salt to 5 ml of distilled water for 30 minutes.); 1g of Airflex 300; 0.8 ml of Igepal CO-530 solution (prepared by mixing 0.03 g of a surfactant Igepal CO-530 with 1 g of distilled water); 0.4 ml of semicarbazide solution (0.1 g of semicarbazide was dissolved in 1g of distilled water); 0.3 ml of  $\text{CaCl}_2$  solution (1.4704 g of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  dissolved in 10 g of distilled water); 0.7 ml of 4-  
20 sulfobenzoate solution (prepared by adding 2.4027g of 4-sulfobenzoic acid, potassium salt to 10 g of distilled water. This solution was stirred while 1.75 ml of 10 N NaOH was used to adjust the pH to 5.0. The solution was then filtered through filter paper.); and 2.375 ml of distilled water. The solution was stirred for 30 minutes. Then, 0.84 ml of enzyme



solution was added and stirred for 15 minutes. This enzyme solution was prepared with 2225  $\mu$  of glucose oxidase, 1918  $\mu$  of horseradish peroxidase, 0.2 g of Crotein spc (Croda Inc. NJ) dissolved in 0.6 g of distilled water.

5 Trinder reagent solution: To 1 g of 2,5 Furandione, polymer with methoxyethene (Gantrez An-169 solution) was added 9 g of distilled water, and the pH was adjusted to 7.0 with 10 N NaOH solution while the solution was stirred. To 0.2 g of this Gantrez solution was added 28.7 mg of N-ethyl-N-(3-sulfopropyl)-3,5-dimethylaniline, sodium salt, monohydrate (MAPS), 7.1 mg of 4-aminoantipyrine, 0.6g of distilled water. The solution  
10 was stirred to dissolve. Then 1.1 ml of ethanol was added drop by drop while stirring.

### **EXAMPLE 3B**

#### **Comparison of Prior Art Glucose Test Strip and Test Strip of the Present Invention**

15 Using Composition C, Trinder reagent solution and the substrate described in Example 3A, the procedure set forth in Example 1D was repeated. The results of the glucose determinations are set forth below in Table III.

**Table III**

20

		Sample 1**	Sample 2	Sample 3**	Sample 4	Sample 5**	Sample 6
		mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose	mg/dL Glucose
		41% Hematocrit	51% Hematocrit	41% Hematocrit	51% Hematocrit	41% Hematocrit	52% Hematocrit
	Prior Art Strip	85	81	164	154	456	433
5	run 1	106	98	146	137	402	382
	run 2	110	102	150	141	409	405
	run 3	107	92	145	143	383	387
	run 4	109	97	151	139	394	400
	Ave.	108	97.5	148	140	397	393.5
10	Norm*	108	102.05	148	149.09	397	414.4

\*\* Samples 2, 4, and 6 of higher hematocrit were prepared from Samples 1, 3, and 5 of lower hematocrit by removing plasma therefrom.

\* Strip values of 51% hematocrit sample were normalized by multiplying by a certain correction factor (One Touch value of 41% hematocrit sample which was used to prepare the 51% hematocrit sample / value of 51% hematocrit sample). Thus, the normalized values of the higher hematocrit samples can be compared easily with those of the lower hematocrit samples.

These results indicate that test strips of the present invention are able to determine the levels of glucose present over a wide range of concentrations. Furthermore, the results indicate that interference due to higher hematocrit levels can be easily compensated for using One Touch™ glucose meters.

In the present example, the oxidized TMB, MAOS and MAP products (the chromophores) all absorb substantially at both 635 nm and 700 nm. Thus, the present invention can be used to detect analytes, i.e., glucose, using such chromophores.

- 5           The invention being thus described, it will be obvious that the same may be varied in many ways. Such variations are not to be regarded as a departure from the spirit and scope of the invention and, all such modifications are intended to be included within the scope of the following claims.

**WHAT IS CLAIMED IS:**

1) A method for indirectly determining the presence of one or more analytes in a multi-component medium comprising:

5 a) reacting said multi-component medium with one or more signal-producing reagents to form one or more signal producing molecules which produce spectra in response to one or more stimuli;

b) correcting for the presence of interfering components in said medium by providing or self-producing a stimulus or stimuli for exciting said multi-  
10 component medium whereby the number of stimuli provided or self-produced is equal to or greater than the sum of the number of said analytes in said multi-component medium and any interfering components therein, provided that when there is more than one analyte or interfering component in said multi-component medium that said stimuli are different from a stimulus for measuring said analyte and no two interfering components produce a  
15 signal that overlap at a stimulus that is used for said corrections;

c) detecting at least a first and second signal of said one or more analytes generated in response to said stimulus; and

d) determining the presence and/or concentration of said one or more analytes by applying a mathematical formula to said at least first and second signal.

20

2) The method according to claim 1, wherein said analyte's presence is detected with a percent coefficient of variance between about 30% to about 50%.

3) The method according to claim 1, wherein said analyte's concentration is detected with a percent coefficient of variance of about 3% to about 7%.

4) The method according to claim 1, wherein said stimulus comprises radiation from the electromagnetic spectrum.

5) The method according to claim 1, wherein said at least first and second signals are detected by exciting said medium at more than one excitation wavelength and measuring corresponding absorbencies at said at least first and second wavelengths  $\lambda_a$  and  $\lambda_b$ , respectively.

6) The method according to claim 5, wherein said presence and/or concentration of said one or more analytes is proportional to the magnitude of a signal (Aa) and is determined according to formula I:

$$\text{Concentration of Analyte} = \text{ExAa}$$

wherein  $Aa = (Aa' - kxBb') / (1 - kc)$  Aa is said true value for said analyte, Aa' is a measurable signal produced in response to a stimulus by said signal producing molecule (Aa) and no more than one of said interfering components (Ba) at  $\lambda_a$ , Bb' is a measurable signal produced in response to said stimulus by said signal producing molecule (Ab) and no more than one of said interfering components (Bb) at  $\lambda_b$ , k is a constant and is a ratio between Bb and Ba and c is the ratio between Aa and Ab.

7) The method of claim 6, wherein  $\lambda_a$  is about 630nm and  $\lambda_b$  is about 700nm.

5 8) The method according to claim 1, wherein said signal producing molecule produces a measurable response having a characteristic ratio of signals at two different wavelengths and/or stimuli.

9) The method according to claim 1, wherein said signal producing molecule produces a signal selected from the group consisting of absorbance, voltage, radioactivity, temperature, current, fluorescence and mixtures thereof.

10

10) An analyte detection system for determining the presence and/or concentration of glucose in a blood sample in contact with a porous substrate having a blood application surface and an analyte detection surface wherein the presence and/or concentration of glucose cannot be directly determined comprising:

15

a) means for introducing said blood sample onto said blood application surface;

b) means for allowing said sample to migrate through said porous substrate to said analyte detection surface, wherein said detection surface includes a glucose-specific reagent and an indicator disposed thereon;

20

c) means for forming a signal producing composition through the reaction of said glucose-specific reagent, said indicator and said glucose present in said blood sample;

d) means for stimulating said sample with one or more stimuli;

e) means for detecting one or more signals produced in response to d, and for correcting for any interfering signals produced by other analytes and/or interfering components, wherein the number of stimuli required is equal to or greater than the sum of the number of analytes in said sample and any interfering components therein, provided  
5 that said stimuli are different from a stimulus for measuring said glucose and no two interfering components produce a signal that overlap at a stimulus that is used for said corrections; and

f) means for determining the presence and/or concentration of said glucose by applying a mathematical formula to said one or more signals.

10

11) The system of claim 10, wherein said porous substrate further comprises a matrix of a hydrophilic polymer, said polymer having pores dispersed throughout said matrix and in communication with said blood application and analyte detection surfaces, wherein the diameter of said pores is sufficiently large to allow red blood cells in said  
15 sample to freely migrate therebetween.

12) The system according to claim 10, wherein the presence of glucose in said blood sample is detected with a percent coefficient of variance between about 30% to about 50%.

20

13) The system according to claim 10, wherein the concentration of glucose in said blood sample is detected with a percent coefficient of variance of about 3% to about 7%.

14) The system according to claim 10, wherein said one or more signals are detected by exciting said blood sample at at least two excitation wavelengths and measuring its absorbance produced at said at least two wavelengths  $\lambda_a$  and  $\lambda_b$ , respectively.

5 15) The system according to claim 14, wherein said presence and/or concentration of glucose is proportional to the magnitude of a signal produced by said glucose (Aa) and is determined according to mathematical formula I:

$$\text{Glucose concentration and/or presence} = ExAa$$

10 wherein  $Aa = (Aa' - kxBb')/(1 - kc)$  and Aa is said true value for said glucose, Aa' is a measurable signal measured at  $\lambda_a$  and produced in response to a single stimulus by said signal producing composition (Aa) and any interfering red blood cells (Ba), Bb' is a measurable signal measured at  $\lambda_b$  and produced in response to said stimulus by said signal producing composition (Ab) and said interfering red blood cells (Bb), k is a constant and is a ratio between Bb and Ba, and c is the ratio between Aa and Ab.

15

16) The system according to claim 14, wherein  $\lambda_a$  is about 630nm and  $\lambda_b$  is about 700nm.

20 17) The system according to claim 10, wherein said signal producing composition produces a measurable response having a characteristic ratio of signals at two different wavelengths.

18) The system according to claim 10, wherein said signal producing



composition produces a signal selected from the group consisting of absorbance, voltage, radioactivity, temperature, current, fluorescence and mixtures thereof.

19) The system according claim 10, wherein said glucose-specific reagent  
5 includes a chromogen.

20) The system according to claim 19, wherein said chromogen is 3,3',5,5'-  
Tetramethylbenzidine.

10 21) The system according to claim 10, wherein said glucose-specific reagent is  
a Trinder reagent.

22) A process for detecting one or more analytes in a multi-component system  
wherein the ability to directly detect said one or more analytes is masked by one or more  
15 interfering components of said multi-component system comprising:

a) forming a signal producing composition by reacting an analyte with  
a reagent;

b) exciting said system with one or more stimuli to produce an  
interference-correction signal from said signal producing composition, whereby the  
20 number of stimuli provided is equal to or greater than the sum of the number of said  
analytes in said multi-component medium and any interfering components therein,  
provided that said stimuli are different from a stimulus for measuring said analyte and no  
two interfering components produce a signal that overlap at a stimulus that is used for said

corrections;

- b) measuring said signal or signals; and
- c) determining the presence and/or concentration of said one or more

analytes by applying a mathematical formula to said signal or signals, wherein said signal  
5 or signals are proportional to said analyte or analytes to be detected said mathematical  
formula being:

$$\text{Concentration and/or Presence of Analyte} = \text{ExAa}$$

wherein  $Aa = (Aa' - kxBb')/(1 - kc)$  and Aa is said true value for said analyte, Aa' is a  
measurable signal produced in response to a single stimulus by said signal producing  
10 composition (Aa) and said no more than one interfering component (Ba) at said first  
detection point, Bb' is a measurable signal produced in response to said stimulus by said  
signal producing composition (Ab) and said no more than one interfering component (Bb)  
at said second detection point, k is a constant and is a ratio between Bb and Ba, and c is the  
ratio between Aa and Ab.

15

23) The process according to claim 22 wherein said analyte's presence is  
detected with a percent coefficient of variance between about 30% to about 50%.

20

24) The process according to claim 22 wherein said analyte's concentration is  
detected with a percent coefficient of variance of about 3% to about 7%.

25) The process according to claim 22, wherein said signal producing  
composition produces a measurable response having a characteristic ratio of signals at two

different stimuli.

5           26)    The process according to claim 22 wherein said signal producing composition produces a signal selected from the group consisting of absorbance, voltage, radioactivity, temperature, current and fluorescence.

          27)    The process according claim 22, wherein said analyte is glucose and said reagent is a glucose-specific reagent.

10           28)    The process according to claim 27, wherein said glucose-specific reagent is a chromogen.

          29)    The process according to claim 28, wherein said chromogen is 3,3',5,5'-Tetramethylbenzidine.

15           30)    The process according to claim 22, wherein said signal producing composition is a Trinder reagent.

          31)    A device for detecting the presence and/or concentration of one or more analytes in a blood sample when said analyte cannot be directly detected comprising:

20           a)    a hydrophilic porous substrate with a blood application surface and an analyte detection surface, said substrate comprising a polymeric matrix having pores dispersed throughout said matrix and in communication with said blood application and

analyte detection surfaces, wherein the diameter of said pores is sufficiently large to allow red blood cells in said sample to freely migrate therebetween;

- b) an analyte-specific reagent and an indicator dispersed on said analyte detection surface for reacting with said analyte to produce an a signal producing composition which, when excited, produces at least two signals at wavelengths ( $\lambda_a$ ) and ( $\lambda_b$ ), respectively to correct for any interfering signal or signals produced by other analytes or interfering components in said sample;
- c) excitation means for exciting said sample on said analyte detection surface;
- d) detection means for detecting said at least two signals, whereby the number of signals produced must be equal to or greater than the sum of the number of said analytes in said blood sample and any interfering components therein, provided that said signals are different from a signal for measuring said analyte, and no two interfering components overlap at a stimulus that is used for said corrections;
- e) means for determining the presence and/or concentration of said analyte by applying mathematical formulas to said at least first and second signals, said formula being:

$$\text{Concentration and/or presence of Analyte to be detected} = \text{ExAa}$$

- wherein  $Aa = (Aa' - kxBb')/(1 - kc)$  and Aa is said true value for said analyte, Aa' is a measurable signal produced at  $\lambda_a$  by said signal producing composition (Aa) and no more than one interfering component from said sample (Ba), Bb' is a measurable signal produced at  $\lambda_b$  by said signal producing composition (Ab) and no more than one

interfering component from said sample ( $B_b$ ),  $k$  is a constant and is a ratio between  $B_b$  and

$B_a$ ;  $c$  is the ratio between  $A_a$  and  $A_b$ ; and

f) display means for displaying said presence and/or concentration of said analyte.

Figure 1

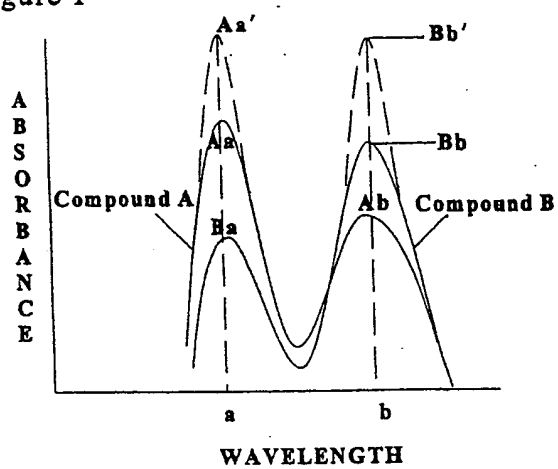
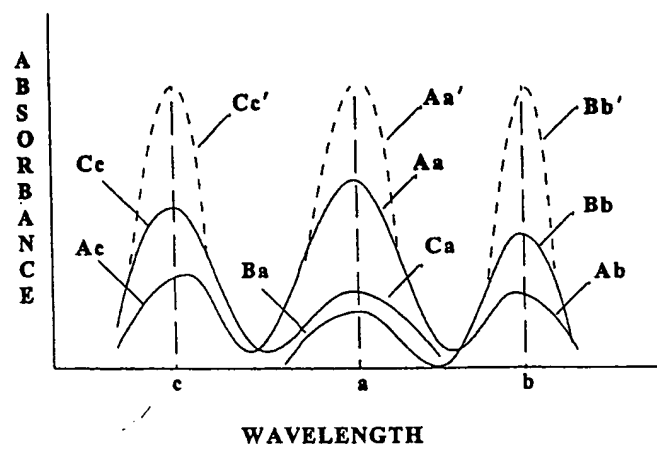


Figure 2



## INTERNATIONAL SEARCH REPORT

 International application No.  
PCT/US98/16980

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) :G01N 21/77, 33/52

US CL :Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 436/46, 93, 94, 95, 164, 166, 169, 172; 422/56, 82.05, 82.08, 82.09

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

Please See Extra Sheet.

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X — Y	J. L. Xu et al, "A New Solution of Simultaneous Equations and its Application in Assay of Compound Preparations" Acta Pharmaceutica Sinica, 1990, Vol. 25, No. 8, pages 626-631, see entire document.	1-6,8-9,22-26 ----- 7,10-21,27-31
X — Y	H.-W. Gao, "Dual-Wavelength Beta-Correction Spectrophotometric Determination of Arsenic in Wastewater with Ethyl Violet" Talanta, 1995, Vol. 42, pages 891-895, see entire document.	1-5,8-9 ----- 10-14,16-21
X — Y	Z. Zhang et al. "A New Calculation Method in Three-Wavelength Spectrophotometry and its Application" Fenxi Huaxue, 1985, Vol. 13, No. 4, pages 262-265, see entire document.	1-5,8-9 ----- 10-14,16-21

☒ Further documents are listed in the continuation of Box C.
 ☐ See patent family annex.

* Special categories of cited documents:	*T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A* document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*B* earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*Z* document member of the same patent family
*O* document referring to an oral disclosure, use, exhibition or other means	
*P* document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

04 NOVEMBER 1998

Date of mailing of the international search report

03 DEC 1998

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## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US98/16980

## C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 4,935,346 A (Phillips et al) 19 June 1990, see entire document.	10-21,31
A	US 4,772,561 A (Genshaw) 20 September 1988.	1-31
A	S. Brewer, "Solving Problems in Analytical Chemistry" 1980, John Wiley & Sons, Inc. (New York), pages 256-285.	1-31
A	N. F. Jaffery et al, "A Spectrophotometric Method for Simultaneous Estimation of Phenytoin and Carbamazepine" Journal of Pharmacological Methods, 1983, Vol. 9, pages 33-39.	1-31

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/US98/16980

## A. CLASSIFICATION OF SUBJECT MATTER:

US CL :

436/46, 93, 94, 95, 164, 166, 169, 172; 422/56, 82.05, 82.08, 82.09

## B. FIELDS SEARCHED

Electronic data bases consulted (Name of data base and where practicable terms used):

STN search in CA file

search terms: multicompon?, multi, multiple, compon?, analyt?, component, interfer?, wavelen?, determin?, measur?, monitor?, estimat?, anal?, detect?, optimiz?, interval, algorit?, function, formula, mathe?, algebra, computat?, complex, reagent, tetramethylbenz?, trin!r, glucose, expression, orthogonal, linear, dual, beer, lambert, correct?, remov?, eliminat?, beta, law, absorpti?, molar, coeffic?, ratio, proprtion, overlap?, spectrophoto?